

Lipid Extraction of Tissues with a Low-Toxicity Solvent

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An improved method for extracting the lipids from tissues consists of the use of hexane:isopropanol, followed by a wash of the extract with aqueous sodium sulfate to remove nonlipid contaminants. This method has a number of advantages over the common usage of chloroform:methanol. The solvents are somewhat less

low enough to permit centrifugation of the homogenate as an alternative to filtra-

of gangliosides.

Biochemists wishing to extract lipids from tissues have in the past paid rather little attention to questions of danger to their health or to the health of people living downstream from their sewage treatment plants. Recent findings from public health studies and animal experiments have implicated a wide variety of organic solvents in the production of tumors, through

hazard. While many solvents have been recommended [for reviews see Refs. (1-3)], the most popular is the chloroform:methanol (CM)¹ system of Folch *et al.* (4) and its various modifications. Chloroform can produce tumors in animals (5) and methanol is well known for its damage to the visual system.

A choice of solvents for lipid extraction should be made on the basis

of several additional factors: volatility (for ready removal later), freedom of form a two-phase system with water (to remove nonlipids), extraction power for undesired components (proteolipid proteins, small molecules), price, range of extraction power for the different lipid classes, and ultra-violet transparency for subsequent column chromatography and monitor-

¹ Abbreviations used: HIP, hexane:isopropanol, usually in the ratio 3:2. CM, chloroform:methanol, usually in the ratio 2:1.

ing in the low wavelength region. Chloroform rates low with respect to impurities (HCl and phosgene), price, and transparency. After examining the various available solvents, we turned to hexane:isopropanol (HIP). Hexane in high concentrations is a known neurotoxin, apparently through metabolism to the 2,5-diketo compound (6), but it seems to be relatively nontoxic in laboratory usage and probably is not a hazard after sewage processing. Until recent years commercial hexane was contaminated with sulfur and aromatic compounds, which tend to polymerize to nonvolatile materials, but contemporary hexane is much purer. Our trials with HIP indicate that it satisfies many of the needs of the lipid chemist and shows significant advantages over other systems.

METHOD AND EVALUATION

Materials. Hexane ("nonspectro") was the product of Burdick & Jackson Laboratories, Muskegon, Michigan. Some work was done with "hexanes," J. T. Baker Analyzed Reagent grade, distilled before use. Isopropanol was the reagent grade product, also redistilled. Radioactive ganglioside G_{M1} made by the galactose oxidase/borohydride method (7), and crude G_{M1} were gifts of Dr. K. Suzuki. Radioactive glucocerebroside and sphingomyelin were prepared here, and labeled cerebroside sulfate was a gift from Dr. Y. Kishimoto.

Thin-layer chromatography was done with commercially coated silica gel plates (EM Laboratories), using chloroform:methanol:water (65:25:4) or hexane:diethyl ether:acetic acid (80:20:1). The plates were sprayed with iodine in methanol, ninhydrin, phospholipid detecting spray (8), or a cupric:phosphoric acid charring spray (9).

Column chromatography was carried out with Silica Gel 60, 230/400 mesh (EM Laboratories).

Lipid extraction technique. Rat or mouse brain was used for all the tests.

the suspension was stirred through a sintered glass Buchner funnel (medium porosity) fitted with a ball joint for use with pressure (10). The homogenizer, funnel, and residue were washed three times with 2-ml portions of HIP, by resuspending the residue each time and letting the solvent

pooled nitrates for at least 1 mm with 12 ml of aqueous sodium sulfate (prepared from 1 g of the anhydrous salt and 15 ml of water). The two layers that formed were each about 18 ml in volume. The lipids were in the upper, hexane-rich layer. No precipitate was visible at the interface.

Caution. Hexane is relatively flammable and ventilation may be needed near the homogenizer.

Evaluation of the HIP extraction method. Most of the evaluation was

(4). One gram of chopped, blended brain was homogenized as above with 18 ml of chloroform:methanol (2:1) and the residue was rinsed as above with 3×2 ml of the same solvent. Washing of the pooled filtrates was done with 4.8 ml of 0.88% KCl.

When a CM filtrate was evaporated to dryness and dissolved in HIP or CM, much denatured proteolipid protein and other insoluble material could be seen. In the case of HIP extracts, the solutions were clear, indicating the absence of proteolipid protein and a reduced amount of nonlipid contamination. This interpretation was confirmed by analyzing the washed extracts for protein by the method of Lowry *et al.* (11), modified for proteolipids (12). The CM extract contained 3.2 mg of protein/g of brain, while none could be detected in the HIP extract.

An attempt to extract the proteolipid protein from the residue remaining from HIP extraction, by treatment with chloroform-methanol-water 10:10:1, yielded only 0.9 mg of protein (0.5 mg in the case of CM residue). Evidently the HIP treatment denatures it, or fails to convert the myelin

HIP extraction is light in color and fluffy.

The HIP extracts apparently contained less of the other nonlipid tissue materials. In an experiment with 1 g of brain, the first CM extract contained 95.2 mg of dry solids while the HIP extract contained 79.9 mg (15.3 mg less). Washing the extracts once with KCl or Na_2SO_4 removed 15.7 mg from CM and 10 mg from HIP. The first CM extract then contained 79.5 mg less material than the CM extract; only about 3.2 mg of this difference

Examination of all four extracts, washed and unwashed, by thin layer chromatography, using 0.1 and 0.3-mg of lipids, revealed no differences in the major bands but distinct differences were visible in the material at the origin and just above. Ninhydrin revealed a distinct band at the origin from CM extracts and a lesser band just above; the unwashed HIP extract also showed these bands, but only faintly. Iodine showed the low bands even more clearly and a trace was now visible even in the washed HIP lipids. Evidently a single wash of the CM lipids was insufficient to remove an appreciable amount of nonlipid material.

The lipids extracted by the two solvent systems were compared after column chromatography. Extraction was carried out as above with 2 g of rat brain and the unwashed extracts were concentrated to dryness with a stream of nitrogen, followed by high vacuum desiccation. Both preparations were treated with 18 ml of 70 mM NaOH in CM (13) for 60 min and soluble nonlipids were removed by partitioning with 4.5 ml of 0.28 M

aqueous acetic acid. The lower layer was washed twice with 10 ml of 0.88% KCl:methanol (1:1). precipitated protein in the CM lipids was removed

were suspended in hexane and applied to a column containing 6 g of

ml of hexane-toluene 1:1 yielded the fatty acid methyl esters (72.7 mg from HIP, 74.3 mg from CM), 150 ml of HIP 95:5 yielded free fatty acids

isopropanol-water 60:80:14 yielded the polar alkali-stable lipids (60.8 and

This study showed that both solvents extracted very similar amounts of most of the lipids, but that the methyl ester yield was slightly higher with CM. This difference was observed in several experiments. Part of the difference must arise from the failure of HIP to extract proteolipid protein, which contains fatty acids that are rapidly liberated by alkaline methanolysis (14).

Thin-layer chromatography of the lipids obtained by column fractionation showed no difference between the two extracts. Gas chromatography of the methyl esters with programmed temperature and poly(diethylene-glycol succinate) columns showed there was no difference in fatty acid distribution. The content in weight percentage of each major fatty acid was 20% 16:0, 20% 18:0, 26.5% 18:1, 3% 20:1, 10% 20:4, 3% 22:4, and 14% 22:6.

The fatty acids missing from the HIP extract were sought in the insoluble protein/nucleic acid residue. This residue was dried, then heated overnight at 100°C in a sealed tube with 8 ml of 4 M methanolic HCl. The fatty acid methyl esters were extracted with 3 × 16 ml of hexane, the esters were purified by chromatography on silica gel as above, and the weights were determined by gas chromatography as above, using an internal standard, tricosanoic acid. The HIP residue contained 3.4 mg and the CM residue contained 1.5 mg (a difference of 1.84 mg). While part of this difference in contents is due to proteolipid protein fatty acids,

extraction as noted below.

The distribution of fatty acids in the proteinaceous residue differed appreciably from that seen in whole brain and the two residues. The HIP residue contained 22% 18:0, 39% 18:0, 13% 18:1, 2% 20:0, 12% 20:4, and 5% 22:6. The CM residue

some of these fatty acids were derived from polyunsaturated fatty acids, which

and are largely removed from the extract by partitioning with aqueous salts. Ganglioside behavior in HIP was tested by evaporating to dryness

(NAN)GalGlc ceramide]. To this was added 0.5 g of chopped brain and the lipids were extracted with HIP or CM. In the case of HIP, only 44% of the tritium was found in the filtrate; in the case of CM, 95% was found. Washing the HIP with Na_2SO_4 solution left only 7% of the ganglioside activity; washing the CM with KCl solution left 33%. Evidently gangliosides are not very soluble in HIP and the aqueous partitioning removes them efficiently. It seems likely that using several portions of water-containing HIP, or raising the proportion of isopropanol would give more

A similar test with labeled glucocerebroside and sphingomyelin showed that the HIP extract, after washing, contained 99.2% and 97.6% of the two lipids. When the aqueous layer was backwashed with hexane:isopropanol 7:2 (which approximates the upper layer in composition), an additional

the upper layer. Evidently the HIP extraction step is efficient for these lipids, but 2–3% of the lipids are lost when nonlipids are removed with aqueous Na_2SO_4 . Presumably the loss on washing is less for the less polar lipids. In any event, the recovery can be made complete by backwashing the aqueous layer. A test with labeled cerebroside sulfate showed that 96.7% of the lipid was retained in the hexane layer after washing.

In searching for a useful washing system for removing nonlipids from

loss but an appreciable amount of KCl entered the hexane-rich phase. We then tried a salt containing a divalent ion, which should reduce hexane solubility while increasing the salting out effect. Only traces of Na_2SO_4 were found to enter the upper layer. However we found that the concentra-

differ in lipid distribution in a useful way.

DISCUSSION

The HIP extraction technique described here has several desirable properties, particularly when it is compared with CM. The HIP extract contains less nonlipid material and, after evaporation to dryness, it can be applied to a chromatographic column or thin-layer plate without fear of

is applied to red blood cells, somewhat less pigment is extracted. If one wishes to examine the proteinaceous residue for other analyses, the homogenate can be centrifuged and the residue washed in the test tube without loss. (However, the use of dry HIP for the washing causes some

material to float; this is prevented by including 2.5% water in the HIP.) When this approach is attempted with CM, one must reduce the density

amount of nonlipid contamination of the lipids and one must add chloroform to carry out the aqueous washing step.

In most cases, the HIP extract can be processed without washing. If it is important to remove nonlipids by washing, this can be done with aqueous Na_2SO_4 . In this event, if complete recovery of the lipids is needed, a backwash of the aqueous layer with HIP 7:2 is necessary. These partitioning steps yield rapid phase separation without interface precipitation; CM extracts sometimes produce an interfacial material during washing.

CM, it is common for investigators to use a two-step solvent addition system, as suggested by Bligh and Dyer (15). In this approach, the ratios of water, chloroform, and methanol are such that there are two liquid phases but extraction of lipid seems nevertheless to be adequate. It is likely that the HIP system could be modified similarly.

It is also possible that the washed HIP extract obtained by our procedure could be applied directly to a silica gel column for the isolation of the more polar lipids. Geurts van Kessel *et al.* (16) have shown that a

of HIP-water mixtures. They monitored the separations with a flow cell spectrophotometer at 206 nm. We have adapted the method for the preparative isolation of lecithin (17) using a higher wavelength to reduce the sensitivity of detection. In this case one can avoid using expensive "spectro" grade solvents.

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