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Surface roughness and characteristics of CAD/CAM zirconia and glass ceramics after combined treatment procedures

Sibel Dikicier* Cumhur Korkmaz and Arzu Atay

Abstract

Background: The roughening of the inner surface of a fixed ceramic restoration is an important factor for the bonding process. The aim of this study is to investigate the effect of combined surface treatments (acid etching, airabrasion and Er: YAG Laser) on surface roughness of CAD/CAM fabricated zirconia (ZrO₂) and lithium-disilicate glass ceramics (LDS).

Methods: Sixty ZrO_2 (Ceramill Zi) and LDS (IPS e.max CAD) specimens, (5 mm in width, 5 mm in length and 1.5 mm in height) were fabricated using CAD/CAM and sintered according to the manufacturer's instructions. All specimens subjected to three surface treatment combinations; etching with 4% hydrofluoric acide (HF), airborne-particle abrasion with 110- μ m alumina (Al $_2O_3$) (AP) and Er:YAG laser (Er:YAG) (Group A—HF + AP; Group B—Er:YAG + AP, and Group C—Er:YAG + HF). Perthometer was used to measure the surface roughness of the specimens before and after the tretments.

Results: Group A presented the highest Ra (LDS 0.81 ± 0.27 and ZrO_2 0.67 ± 0.21 after treatment) and Group C the lowest (LDS 0.45 ± 0.13 and ZrO_2 0.26 ± 0.07 , after treatment). Compared with before treatment, the Ra were significantly different only in Group A both ZrO_2 and LDS after treatment (p < 0.05). Qualitative SEM images suggested the surface topography of the ZrO_2 was smoother than the LDS. Less surface changes were observed in the Er:YAG combined procedures than HF + AP.

Conclusions: HF + AP was significantly successful in modifying the ceramic surface. Er:YAG did not sufficiently promote the surface topography, even if combined with any other treatments. Overall, surface tretments on ZrO_2 not easier than LDS.

Keywords: Surface property, Dental air-abrasion, Dental acid etching, Er-YAG laser

Introduction

Superior esthetic characteristics of all-ceramic restorations made them highly popular through the last decade. With the elimination of metal infrastructures, optimal distribution of reflected light and translucency is achieved which leads to a highly esthetic appearance and simulating the natural appearance of natural tooth [1, 2]. With these advantages all-ceramic restorations are indicated for fixed prostheses such as ceramic inlay/onlay restorations, partial/full crowns and bridges.

Fort the aim of the achieve better prosthetic results, it has developed to the use of restorations produced by the CAD/CAM system (computer-aided design/computer-aided manufacturing), which is an extensive technique in last decades. This technique allowed to manufacture the fixed restorations in a single session

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with excellent accuracy and adaptation [3, 4], using industrially manufactured ceramic blocks [5, 6]. Materials with different compositions and microstructures are available for CAD/CAM, such as lithium disilicate glass—ceramics and zirconium-oxide based polycrystalline ceramics which were most popular on the dental market [6]. Although there were many recent studies, these ceramics were considered in the present study due to their widespread clinical option.

Glass–ceramics composed of leucite or lithium disilicate as basic crystalline structure have become preferred material due to their advanced physical, chemical, and mechanical properties [7]. Lithium disilicate glass–ceramics (LDS) based on SiO₂–Li₂O materials system are particularly commercially successful in dental applications. LDS are processed into full-contour restorations for inlays/onlays, veneers, crowns, and fixed partial dentures either by heat-pressing techniques or computer-aided design/computer-aided manufacturing (CAD/CAM) [7, 8]. The CAD/CAM method does not require multiple firings and the blocks have several advantages, such as fast milling and increased fracture resistance [9].

When yttria-stabilized tetragonal zirconia polycrystal (YTZP) is subject to thermomechanical factors, transformation of its tetragonal to monoclinic phase occurs [10]. This transformation-toughening property is responsible for its high fracture resistance [11] making zirconium oxide suitable for use as a framework for fixed prostheses, resin bonded fixed prostheses, and dental implant abutments [12, 13].

Bonding procedure is crucial for clinical survival rates for all-ceramic restorations. The diffusion of monomers into the demineralized dentine matrix, followed by polymerization, assists the micromechanical connection over hybrid layer formation [14]. Likewise, the internal surface of the ceramic restoration must be modified to manage the micromechanical connection between the ceramic and the resin cement. A number of techniques had been reported to enhance the bond strength between luting cement and ceramic [2, 15]. Etching with hydrofloric acid (HF) provides a well established bonding between resin cements and lithium disilicate glass ceramics which is a popular synthetic glass ceramic with the higher results in esthetic and mechanical characteristics [16, 17]. The microstructure of this ceramic changes by dissociation of one of the glassy phases of ceramic [18, 19]. This phase is fused rather to form an convenient surface structure for bonding [20–22]. Dissolving of the glassy phase exposes lithium disilicate crystals which shows as retentive characters [23]. Additionally, silane primers can provide a chemical bond between resin and glass ceramic [24, 25].

It was reported that hydrofloric acid application for 1–3 min provides successful results in terms of adhesive retention, usually in concentrations ranging from 2.5 to 10% [26–28]. However, some researchers have centered their studies in investigating an alternative surface procedure for glass ceramic, by obtaining better adhesion [29]. In recent years, it was reported that methods such as sandblasting, or laser irradiation for surface treatments of glass ceramics may also provide an optimal adhesion [2, 30, 31].

Zirconium dioxide (ZrO_2) cannot be roughed enough with acid etching due to absence of a glassy phase [10]. Another surface treatment suggested before bonding for ZrO_2 is airborne-particle abrasion (AP) with aluminum oxide which creates an irregular topography with expanding surface area thereby increasing the bond strength of resin to ceramic [32].

Although AP is known to be effective method to surface roughness of ZrO₂, some reports discuss about airborne particle abrasion and the possible long-term adverse effect of external erosion on the strength of ZrO₂ [33, 34]. Tribochemical silica coating and/or laser treatment has been presented as an option to airborne particle abrasion, in a try to advance the surface conditions of ZrO₂. The laser application as ceramic surface treatment is still a disputable subject in dentistry. For dentistry practices especially including surface conditioning on ZrO₂ for obtaining the best bonding strength, erbium:yttrium aluminum garnet (Er:YAG) laser is recommended [31]. Previous in-vitro studies reported that various laser types can be used effectively for modifying the microstructural characteristics of ZrO₂ [35–37].

Many surface treatment methods are valid to develop an efficient bonding for ceramic surfaces. There was not sufficient information about the combination of both surface treatments which to create better roughness on different all-ceramic restorations. Thus, the present study aimed to evaluate the effect of varying combined surface treatments such as: HF with AP, HF with Er:YAG, and AP with Er:YAG, on surface conditions of $\rm ZrO_2$ and lithium-disilicate glass ceramic. The rationale for testing CAD/CAM ceramic surfaces is based on the clinical situation where adhesive cementation is needed. The null hypothesis were tested that various surface treatment combinations would not affect the surface rougness on zirconia and lithium disilicate ceramics.

Materials and methods

Sample preparation and surface treatments

Thirty $(5 \times 5 \times 1.5 \text{ mm})$ pre-sintered zirconium oxide (ZrO_2) (Ceramill Zi; Amann Girrbach) and thirty $(5 \times 5 \times 1.5 \text{ mm})$ lithium disilicate glass ceramic (LDS) (IPS e.max CAD, Ivoclar-Vivadent) block-shaped samples

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were fabricated according to the manufacturer's instructions from CAD/CAM machineable blocks. The surfaces of the samples were finished with a 600-1200 grit metallographic paper with a polishing machine (Labpol 8-12, Extec). and then crystallized or sintered according to the manufacturer's instructions (with a temperature of 1600 °C for ZrO2 and 770 °C LDS). Glazing was not applied since it was purposed to evaluate the untreated inner surface of the ceramic. All specimens were divided into two groups according to the ceramic material (n=30/ceramic material), where each group was further subdivided into three subgroups (n=10/subgroup)according to the surface treatment type (Table 1). Samples in Subgroup A were etched with 4% hydrofluoric acid gel (Porcelain Etch Hydrofluoric Acid, Ultra-dent Products Inc.) for thirty seconds and sandblasted with air-abrasion (110 μm Al₂O₃) (Korox, 110#46,014; BEGO), in subgroup B; were treated with Er:YAG laser and sandblasted with air-abrasion (110 µm Al₂O₃), and in subgroup C; were treated with Er:YAG laser and etched with 4% hydrofluoric acid gel. After the airborne particle abrasion, samples were washed with drinking water for 1 min and ultrasonically cleaned in a water bath for 10 min and air dried.

Surface analysis and SEM procedure

Surface roughness of the samples was evaluated with Perthomether M2 (Mahr Co.) measuring device to scan the surface roughness with a microneedle, utilizing the surface roughness parameter (Ra). Three measurements were performed on the surface of each sample in following directions with a cutoff value of 0.25 mm (λ c) and a speed of 0.1 mm/s, longitudinal, transversal, and oblique [38]. The surface profile was recorded and the mean Ra expressed in μ m was determined.

The superficial topography was observed using scanning electron microscope (JSM-6 6400, JEOL). For this procedure, the samples were coated with a gold–palladium alloy spray and observed to evaluate the surface model. Photographs were taken at a magnification of $500 \times /1000 \times$ and used for comparison in surface smoothness.

Statistical analysis

A descriptive analysis of the roughness data (Ra) were evaluated by using SPSS (Statistical Package for the Social Sciences, version 13.5, SPSS Inc.) to determine the mean and standard deviations. Comparisons within the groups for differences in surface roughness before and after treatment were performed using Wilcoxon signed rank test. The effects of combined surface treatment procedures on between ceramic groups were performed using one-way ANOVA. p value was set at < 0.05.

Results

The comparisons are presented in Table 2. According to surface analysis, among both Ra values, Group A (AP combined HF) presented a statistical difference (p < 0.05), in both $\rm ZrO_2$ and LDS specimens (Ra for $\rm ZrO_2$, 0.26/0.67, Ra for LDS, 0.34/0.81, before / after treatment, respectively). Group A also led to statistically significant higher Ra values in comparison to Group B (Er:YAG combined AP) and/or Group C (Er:YAG combined HF) (p < 0.05). None of the Er:YAG combination treatments of surface treatment groups (Group B and C) did not showed statistically difference in all surface treatment group specimens.

Regardless of the type of surface treatment applied on the ceramic, the Ra values of LDS was slightly higher than that of $\rm ZrO_2$. It must be emphasized that the Ra values between $\rm ZrO_2$ and LDS specimens were similar within both treatment groups. Additionally we may indicate that all treatment protocols were better than the untreated specimens.

SEM micrographs (500×/1000× magnification) showed different surface characteristics of specimens subjected to studied procedures. It emphasized that both experimental ceramic types presented the different topographical pattern before and after treatment and that HF and AP procedure modifies the materials surface attributes its external pattern (Figs. 1, 2), while Er:YAG combined conditions appears to few relate in such outcome (Figs. 3, 4, 5, 6).

Table 1 Test groups and surface treatments

Group	Treatment protocol				
Group A	Etched with 4% hydrofluoric acid gel for 30 s, rinsed for 1 m, air dried for 10 s $+$ and sandblasted with air-abrasion (110 μ m Al ₂ O ₃) for 20 s, with 4 bar pressure, distance of 10 mm				
Group B	Er:YAG laser applied energy level was 150 mJ with 10-Hz frequency for 45 s, pulse width 300 μ S + and sandblasted with air-abrasion (110 μ m Al $_2$ O $_3$) for 20 s, with 4 bar pressure, distance of 10 mm				
Group C	Er:YAG laser applied energy level was 150 mJ with 10-Hz frequency for 45 s., pulse width 300 μ S + and etched with 4% hydrofluoric acid gel for 30 s, rinsed for 1 m, air dried for 10 s				

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Table 2 Surface roughness values (Ra) of various surface treated all-ceramics

Group	Treatment protocol	Material	Mean(SD) Ra (μm)		p*
			Before treatment	After treatment	
Group A (n = 20)	HF+AP	Zirconium oxide	0.2(0.09) ^a	0.6(0.01) ^b	< 0.05
		Lithium disilicate	0.3(0.05) ^a	0.8(0.07) ^c	< 0.05
Group B ($n = 20$)	Er:YAG + AP	Zirconium oxide	0.2(0.05) ^a	0.50(0.07) ^a	> 0.05
		Lithium disilicate	0.36(0.09) ^a	0.4(0.09) ^a	> 0.05
Group C ($n = 20$)	Er:YAG + HF	Zirconium oxide	0.2(0.07) ^a	0.35(0.07) ^a	> 0.05
		Lithium disilicate	0.23(0.09) ^a	0.55(0.13) ^a	> 0.05

^{*}Wilcoxon signed ranks test (p < 0.05)

Within any column and line means with the same superscript letters are not significantly different (p > 0.05). Different letters indicate a significantly difference among the groups (p < 0.05)

HF hydrofluoric acid etching, AP airborne particle abrasion, Er:YAG erbium:yttrium aluminum garnet laser treatment

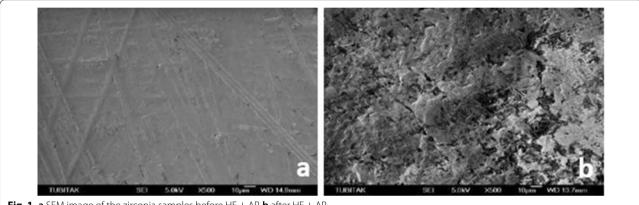


Fig. 1 a SEM image of the zirconia samples before HF + AP, **b** after HF + AP

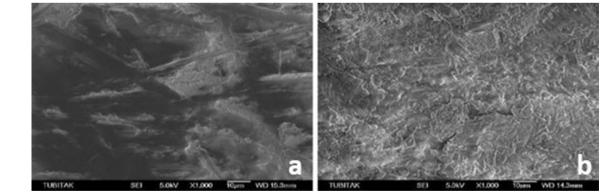


Fig. 2 a SEM image of the zirconia samples before Er:YAG + AP, b after Er:YAG + AP

Discussion

This study investigated the surface parameters in LDS and $\rm ZrO_2$ all-ceramics by different surface treatments. The experimental design used for the etching, airborne particle abrasion and Er:YAG laser treatments were selected based on previous studies [23, 32, 39]. Surface

roughness is one major perspective that describes the efficiency of pre-treatment procedures. For the analysis of the pretreated and treated surfaces of the ceramic specimens, Ra values was used as in many studies [40, 41], but there is no ideal clinically relevant amount of roughness known so far. The findings of this study

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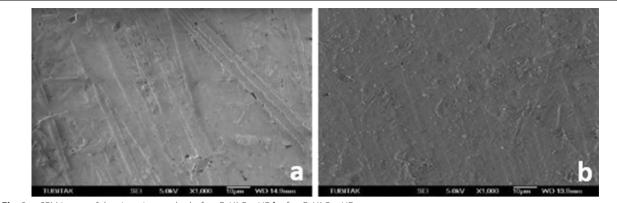


Fig. 3 a SEM image of the zirconia samples before Er:YAG+HF, b after Er:YAG+HF

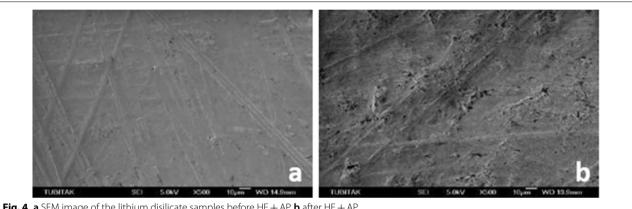


Fig. 4 a SEM image of the lithium disilicate samples before HF + AP, **b** after HF + AP

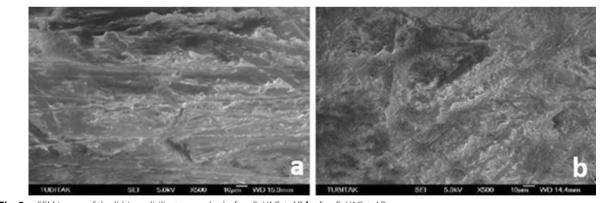
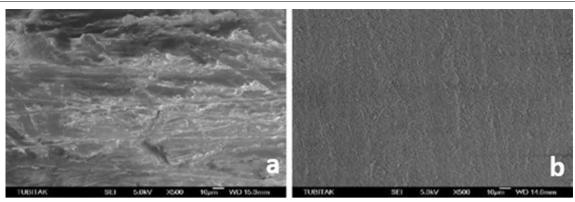


Fig. 5 a SEM image of the lithium disilicate samples before Er:YAG + AP, b after Er:YAG + AP

required the partly rejection of the null hypothesis because changes observed in the surface topography of all-ceramics after surface treatment combinations. However, the proper selection of the surface treatment seemed to be a more important factor relation with ceramic type.

To achieve effective bonding between tooth surface and ceramic, mechanical retention, by surface roughening and microchemical connection with a silane agent are essential. The authors considered that topographic differences of the surface after etching, airborne particle abrasion and/or Er:YAG laser irradiation may have a Dikicier et al. BMC Oral Health (2022) 22:524 Page 6 of 9



 $\textbf{Fig. 6} \ \ \textbf{a} \ \textbf{SEM} \ \textbf{image} \ \textbf{of the lithium disilicate samples before Er:YAG+HF}, \ \textbf{b} \ \textbf{after Er:YAG+HF} \ \\ \textbf{or all the lithium disilicate samples} \ \ \textbf{before Er:YAG+HF}, \ \ \textbf{b} \ \textbf{after Er:YAG+HF} \ \ \textbf{or all the lithium disilicate} \ \ \ \textbf{or all the lithium disilicate} \ \ \textbf{or all the lithium distilicate} \ \ \textbf{or all the lithium distilicate} \ \ \textbf{or all the lithium distilicate} \ \ \ \textbf{or all the lithium distilicate} \ \ \ \textbf{or all the lithium distilicate} \ \ \ \ \textbf{or all the lithium distilicate} \ \ \ \textbf{or all the lithium d$

great effect on adhesion strength. In the present study, among the study groups, statistically significant difference occurs between before and after surface treatment in Group A (HF+AP) when compared with the Group B (Er:YAG+AP) and C (Er:YAG+HF). Furthermore, Group B and C showed similar Ra values, in both ceramic types.

Various surface procedures have been valid to evolve suitable bonding surface to $\rm ZrO_2$ ceramic; such as airborne-particle abrasion, acid etching, tribochemical silica application [42, 43]. AP performed a prepared microretentive $\rm ZrO_2$ surface, increased the adhesion capacity, and improved the surface tension and wettability, hereby increasing the creation of cement-ceramic micromechanical connection [44]. Kim et al. [23] tested $\rm ZrO_2$ ceramics and found that airborne-particle abrasion or acid etching alone few affected to ensure dependable bond strength between resin and $\rm ZrO_2$ ceramics. Additionally, Anand et al. [45] supported the view that conventional methods did not obtain clinically sufficient bond strength values. These results encouraged our study to produce another prebonding surface treatment combinations.

In dental ceramics as LDS, containing glass particles, the surface roughness can be formed with HF for acceptable bonding, while the surface of $\rm ZrO_2$ ceramics is without glass phase; HF does not show any significant increase on bond strength [45]. Therefore, HF was showed a chemical benefit rather than mechanical benefit because previous studies demonstrated it ineffective at surface conditioning in $\rm ZrO_2$ ceramics [1, 2, 46]. In this study, while HF combination treatments indicated higher results for LDS ceramics, it also improved the surface characteristics of $\rm ZrO_2$ ceramic. This effect on $\rm ZrO_2$ was probably caused by the AP stage in the process.

The main structure of IPS e.max Press glass ceramic is shaped by prolonged glass crystals of LDS. Another phase is formed of lithium orthophosphate and a glass matrix encloses both crystalline structures. HF might reshape the glass and crystalline phase in this way composing rougness within the LDS crystals. In addition, the high level roughness improves the surface energy and the connection between the adhesive bonding and silane, thus supporting a micromechanical retention at the ceramicresin interface [47]. The present study showed that 4% HF applied for 20 s combined with AP on the IPS e.max Press glass ceramic is suitable on the glass structure and therefore creates an irregular surface sufficient for bonding. Nevertheless, it should be focused that HF application on glass ceramics, remains as a primary important step on silanization procedures of glass—ceramics.

Although AP, not recommended by the manufacturers for LDS, few studies were presented in the literature on the surface roughness with AP [47, 48]. Gorman et al. [48] reported that AP procedure after etching did not damaged to the etched condition of the ceramic surface, while some abrasive particles may have embedded in the surface. Other study stated that of AP showed the creation of predictable microretentive grooves, but HF formed a microporous surface on SEM images [47]. However, it should be noted that, in the present study, in all combined surface treatment procedures, while the etching phase is more effective on surface roughness for LDS, airborne particle abrasion phase is more effective for ZrO₂.

Many studies were focused that the AP procedure showed significantly higher Ra values on ZrO_2 [39, 49]. It was also indicated that to avoid injuring ZrO_2 and meantime provide optimal bond strength, AP should perform with proper particle size and jet pressure according to manufacturers recommedation.

Several researches have tested the surface treatment of $\rm ZrO_2$ to bonding mechanism with Er:YAG laser. Kunt et al. [31] reported no significant differences in surface roughness after laser treatment on zirconia surfaces, only $\rm CO_2$ laser irradiation technique were found successfully

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and recommended as an alternate surface treatment to AP for ZrO_2 .

Our results showed that the Er:YAG/HF combination slightly changed the ZrO₂ surface with the formation of limited number of micro-porousites. Nevertheless, Er:YAG/AP combined treatment is more effective due to the possible efficiency of AP phase on ZrO₂. Zeidan et al. [38] indicated that there is a significant relation between the Er:YAG laser power capacity and the enhance of surface morphology of zirconia. It was considered that the effect of higher laser power on the ceramic surface revealed increased surface rougness without ceramic loss. The laser energy was selected as 150 mJ in our study, according to previous research and the potency of these factors was evaluated, and the results appeared similar to other surface treatments tested [32].

The tested LDS ceramic is the most preferable high translucent material in prosthodontics, and therefore there are a lot of informations about the surface parameters of this ceramic in the literature [47, 48, 50]. Laser irradiation is not commonly used to surface traetment for LDS. For this reason, there are few data about Er:YAG laser treatment on LDS and present study indicated that the Er:YAG treatment combinations is not as significantly successful as HF combinations on LDS surface. Our data supports a similar performance of the LDS to the ones that have been evaluated by previous studies [29, 51]. Nevertheless, more clinical and in vitro studies are necessary about surface parameters of dental ceramics, since there is still no sufficient data about effective laser type and applications modes.

In the present study, HF+AP combination was found to be effective technique for both ceramic types tested. Moreover, Cervino et al. [52] reported that sandblasting and acid-etching combination was a safe and successful method to modify the titanium dental implant surfaces. It may consider that HF+AP method with varying concentration and particle sizes is effective on metal surfaces in addition to ceramic surfaces. Future studies with dental material-comparative studies on this surface treatment combination may contribute to the clinic and manufacturing processes.

With the disadvantages of ceramic brittleness and new developments in adhesion technology, new materials such as, polyether ether ketone (PEEK) which is a thermoplastic resin polimer, and glass-fiber blocks, which is reinforced resin composite, may become popular as a framework material in prosthetic restorations [53]. It is also advantages that they can be produced with CAD/CAM technique and surface treatment can be applied.

According to the results of our study, all ceramic specimens demonstrated irregular surface after combined surface treatments applied. However, it may be stated

that surface roughness is only a portion of the adhesion mechanism. Study limitations include the reality that are the comparison of alone and combined surface procedures together and testing with the bonding process. Another potential limitation of this in-vitro study is that the clinical situation cannot be completely represented. Therefore, further extensive in vitro and/or in vivo studies are necessary which consist adhesion procedure to approve the results of this in vitro study.

Conclusion

Within the limitations of this in vitro study the following conclusions can be drawn:

- Only the HF + AP combined treatment succeeded on surface rougness of ZrO₂ and LDS ceramic surfaces.
- Er:YAG+HF and/or Er:YAG+AP combined treatments did not significantly increase the surface rougness of the both ceramics tested.
- LDS showed higher Ra values than ZrO₂ regardless of the surface procedure.
- The use of 4% HF acid etching+AP with 110 µm Al₂O₃ resulted in significantly higher surface roughness on to both LDS ZrO₂ ceramics. These results may indicate improved bond strength.
- Conventional techniques such as HF and AP, alone or combined, still appeared to assume a more effective role in the surface treatment of the related ceramic types. Thus, the use of these techniques may bring successful bonding benefits to restorations clinically.

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Author contributions

SD, and AA designed and planned the study; SD, AA and CK prepared the testing procedure; SD and CK performed the data analysis; SD and AA wrote the manuscript, SD, CK and AA reviewed and approved the final version of the manuscript. All authors read and approved the final manuscript.

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Availability of data and materials

All data generated or analysed during this study are included in this published

Declarations

Ethics approval and consent to participate

Not applicable.

Consent for publication

Not applicable.

Competing interests

The authors declare that they have no competing interests.

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