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In situ dynamic tracking of heterogeneous nanocatalytic processes by shell-isolated nanoparticle-enhanced Raman spectroscopy

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Surface molecular information acquired *in situ* from a catalytic process can greatly promote the rational design of highly efficient catalysts by revealing structure-activity relationships and reaction mechanisms. Raman spectroscopy can provide this rich structural information, but normal Raman is not sensitive enough to detect trace active species adsorbed on the surface of catalysts. Here we develop a general method for *in situ* monitoring of heterogeneous catalytic processes through shell-isolated nanoparticle-enhanced Raman spectroscopy (SHINERS) satellite nanocomposites (Au-core silica-shell nanocatalyst-satellite structures), which are stable and have extremely high surface Raman sensitivity. By combining operando SHINERS with density functional theory calculations, we identify the working mechanisms for CO oxidation over PtFe and Pd nanocatalysts, which are typical low- and high-temperature catalysts, respectively. Active species, such as surface oxides, superoxide/peroxide species and Pd-C/Pt-C bonds are directly observed during the reactions. We demonstrate that *in situ* SHINERS can provide a deep understanding of the fundamental concepts of catalysis.

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• de•. a di g•. c e-acii ela i•• hi• a d eaci• echa i i f ig i ca. i .a ce i .he a i al de ig f highl ef cie. ca al $\bullet a^{15}$. Af $e \bullet e$ del • • . e • b b . h • face a d i ee i g ___k .he e ical cio i a ⁶ ¹⁶, ch g o ha beo ade i .hi a ea. H ..., e e, challo go a cia ed ..., i.h. he e $a \cdot i \cdot f$ del $\bullet ... i a l ca a l \bullet a e a i \circ S$. The, ... he de el $\bullet ... f$ e. f in situ face a al i ech i a ha ca be ed. . d cata ccige ac ical ca al • • • highl eac i 🖬 de i able. F e a le, e in situ . ech i e i cl dig e i e al. a i i elec i c c , high- e e (STM), a bio . • ca• i g. •• eli g ic • c e e X-a h elec ec c (XPS) and X-a ab .i. • ec • c ha e beo a lied. •. d • a caal•. • a d c • dii • • , • ch a • face c • ii • • , che ical • . a • , h l gia a d c di a i • • i • • $a^{17,18}$. The e_{--} k ha e g eal i ed • de•. a di g f ca al•i.

O e f.he e. ...videl e ed e face a alei.ech i e i in situ i f a ed (IR) • ec • c , which ca ide ich if air ab . (e. face. c^{-19} , ac^{-20} ad echa i \bullet^{21} . H $__{v}e$ e, i i e dif c l f in situ eac i 🛛 egi f.he ec . L c . a ., he l $a_{max}a$ be egi i i acca ible b Ra a ec c 22 . Thi .ech i e ha bea • ed. ide. if i.e edia e • ecie • ch a e al-C b • d go ecio, hdlg a de face ido²³²⁵, o e al, al Ra^a ecc i ecii e aci e b. i go e al, i . No ii e gh. 1 ace a 1 f face ecie 1 e al 0 ca al ... S face-o ha ced Ra a .ca.e i g (SERS) ha beo led.cic o..hiliiaio,aiffeoa.eodo e e \mathbf{e} , \mathbf{i} , \mathbf{e} , \mathbf{i} , $\mathbf{$ i de ec • i gle lec la 26 29 . Ne e . hela•, • l a fe., • e al

(i a il A, Ag a d C) \dots i b a o. c edo faco ide a la ge SERS effec³⁰. Tho, o i b g ca al .ic coo o cc b g e .he e ab e ab a gea challo ge.

i o.ed a el.echiek Reco.1, g hell-i la ed a a icle-o ha ced Ra a e ec ec 'SHINERS'³¹, ", hich ha beo ega ded a a i e. go e a i ad a ced e ec ec '³². J SHINERS, ha ic A • a a icle (Ra a • ig al a li e ·) a e c a ed ... ih i h le-f ee • ilica • hell , ha e • . . he f *i*.eacig ...i-h a al . ical . a ge . he che ical • i • • .. The ilica • hell a e . hi o gh (j • . a c le f • a e • •) . ha . he • ha ced elec ag e ic / eld ge e a ed a . he face f. he c e e d be d he face f he hell. SHINERS ha b ko helege.a dige face a eial a d hlg i ci le, a e ial a d a h l g. L ha bea \dots idel e d i elec che i. 31,33 35 , life cia ce³⁶, he a e ial a d e ci c d c i d c. 31,37 , a la a e g c. age a d •e ic•d c c • e • i • ^{38,39}, a d dail life⁴⁰.

b .hi $\dots - k$, \dots

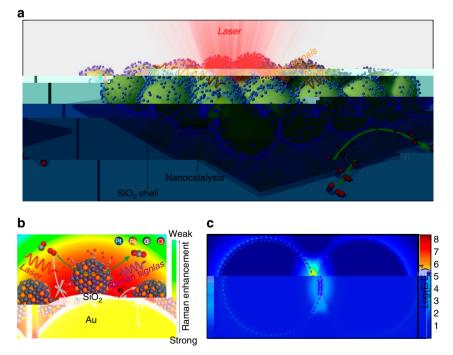


Figure 1 | Schematic illustration of a SHINERS-satellite study of a nanocatalytic process. (a) The Au-core silica-shell nanocatalyst-satellite architecture of SHIN-enhanced Raman spectroscopy (SHINERS)-satellite structures, and the mechanism for CO oxidation over PtFe bimetallic nanocatalysts revealed by our SHINERS-satellite method. The pinhole-free silica shell prevents the Raman signal amplifier from interfering with the system under study. (b) Schematic illustration of CO oxidation on PtFe. The blue, orange, grey and red spheres represent Pt, Fe, C and O atoms, respectively. (c) A 3D-FDTD simulation for a pair of Pt-on-shell isolated nanoparticle (SHIN) nanocomposite structures.

Results

Raman enhancement of SHINERS-satellite nanocomposites. Is SHINERS, ligh - i d ced l cali ede face la •• ield a ea f • ha ced elec ag e ic / eld•. • g h a . h•• face f. he A c e, a d. he elec ag e ic / eld•. • g h deca• e ••.iallvih di. a cef . h•• face f. he A c e de • dieg••ea -/ eld cha ac e i.ia^{31,41}. Th•, ...ve do ig ed a c e • hell• a elli e • a c • i e•. c e ...vih . h••• a ca al••• ...h•• ilica • hell, j•. a c le f• a e e• a...va f ...he c e. Is ...hi ...va, Ra a• ig al f eaci• i.e.

The A -c e ilica hell a ca al . a elli e a chi ec e i • h Fig. 1a. The SHINERS • a elli e . c e i e a ed b caiga A a a icle "i ha lahi e i h le-fee ilica • hell, he addi g he a ca al . . he face f he hell. The i h le-f ee ilica hell e e a ad .i f lec le f .he che ical o i o . o .he A c e, a d i al e e a a i e acii be ... e e . he a ca al a a d .he A c e (Fig. 1b). The, if ai ab . he e go e e ca al .ic eaci ca ca be baied with . ik fi.e fe a ce b . he Ra a \bullet ig al a li/e. T $c \bullet$ / . ha . hi \bullet . c $e = \frac{1}{\sqrt{2}}$ ill effect i el \bullet ha ce Ra a \bullet ig al f \bullet ecia \bullet . he \bullet a ca-.al .a, we delled elec ag e ic eld . o g h i g a 3D-FDTD eh di which he e fec l a ched la e b da $c \cdot di i \cdot \cdot \cdot \cdot \cdot \cdot e e a da . e d. a i d \cdot \cdot - h \cdot i c a e e c i \cdot \cdot \cdot \cdot ^{42}$. The •. c e eleced f dellig _wa a 120 • A c e _wi h a 2 • ilica helle . i g 2 P. • a ca al • . The dielec ic focioo fA a dP. ...ee.ako f a li-cefcio./..ig egi fe, a di a il high elec ag e ic / eld , o g h (a - called h , .') cc i , he a a .icle a a .icle jecie ..., her a ai f SHINERSeaelliee. c e ae ill i a edi h a 633. la e (. he . . al-'eld. ca. e ed-'eld • ce ac ed a a li ea l la i ed ligh • al. . he ai). The

lec la • .he • a ca al • e l ca ed • .he j • c i • be ... e^{-6} .he e^{-6} hell · i la ed • a a .icla (SHIN) ... e^{-1} ld be • ha ced b . 8 de f ag i de, ... e^{-6} hich i • . • g • gh f l a • • i i e de ec i • f • face • ecia ³³.

Synthesis of SHINERS-satellite nanocomposites. The a caale. a ellie a e added. he face f.he SHIN b a elfa e bl ca. The .he f SHIN i da c ibed i de ail el e., he e e ce f ci a e (S le e a Fig. 1). The P. Fe bi e allic a ca ale., ..., hich i ..., idel ed f ge ed c i eacie i f el cell ⁴⁴ a d ca al. ic ida i ⁴, ..., a elec ed a a e e e a ai e ca ale. ..., ih ..., hich i e l e he ..., he f SHINERS a ellie a ca ale. ..., ih ..., hich i e l e he ..., he i f SHINERS a ellie a ca ale. ..., c a. The ..., he i a d cha ace i a i f.he a ca ale. ..., ee e di ed . c ea e a ii el cha ged face (S le e a Fig 1 3), a d i ed ..., ih he ega i el cha ged SHIN (S le e a Fig. 1). achie e he elf-a e bl ca. Fig e 2a clea l h ..., ha he e lig a a.icle ha e ac e hell a ellie .ie ed a chi ec e. Thie. c e i f.he de e... a ed b .he elec e i c c (HR-TEM) i age i S le e a Fig. 4. J a c. l e e i e., di ed P. Fe a ca ale. ..., ee e added. a l f. f SHIN. The did . adhe e. .he ilica face, de e... ai g .ha di ca i f .he P. Fe a ca ale. . c ea e a ii el cha ged face i c ii cal f SHINERS a elli e ... a Fig. 5).

The ghatelet end of the end of t

a .he . a f • a ca al ... - SHIN •. c a
(S le •.a Fig. 6). Fig e 2c • h ... - HR-TEM i aga f
• e allic P. a d Pd• a ca al •., bi e allic P. Pd a d P. Fe
• a ca al •., A @P. Fe c e • hell •. c •, PdFeC • a c ba, a d CeO₂ a d Fe₂O₃ e al ida • SHIN. The i e a d
h 1 g f.ha e • a ca al •. A ge e • cha ged b .he elfa e bl ca•, a d all f.he ... e e • cha ged b .he elfa face f.he SHIN. Th•, i i fea ible. •. d •. c eaci i ela i • hi • a d eaci • echa i • fhe e go e •
caal.ic ca• a b SHINERS af e ca ef ll a i la i g.he
c • ii •, •. c e, h l g a d• i e f.he• a caal •.

• a elli • .

CO oxidation over PtFe bimetallic nanocatalysts. SHINERSa ellie. c $a_{ev}ee he edi ead . dia f CO$ idai e P.-g a caale. Thi ca i f geai .a cei f da e.al eea ch, a d ha bee ec g i ed aa be ch a k e e f he e ge e caale i⁴⁶. I i ali .a. f acical a licai e ch a e i e.al.eci ⁴⁷ a d he d ci f l a e h d ge ⁴⁸.The SHINERS a ellie eh d ..., ed. e a i e.heelai e hi be ..., eea d aci i f P.Fe a caale. .Rece. . dia ha e h ..., ha P.Fe bi e allice a caale. a ehighl aci e ... he idai f CO (ef 45,49). DFT calc lai ei dicae .he high aci i e l f ef cie. aci ai fga e O₂ lec le b .he c di ai el e a a ed Feco. e i P.Fe (ef. 45). Di ec ec c c i c e ido ce i e.illeeded. e.hi echa i . The in situ SHINERS a ellie. a eg i. d ced ab e ca e eal .he effec f.he fe. co. e ... he ef a ce f CO idai.The alo ce ... ae f Fe i .he P.Fe a caale. ...,

dee ied b XPS (geor daa is i .hel...e a. f Fig. 3a). F efe e ce, XPS ec a f Fe i diffe e. ida i 🛛 • ...eig (ble, ed ad black c a i .he e a. f Fig. 3a). Fe f il e ed. . he a he e a_{ra} idi ed a_{ra} hile. he b lk a e ial e ai ed i e allic f . The g ed da a i a i Fig. 3a h a_{∇} -a Fe 2 $_{3/2}$ eak a ~710.4 eV. We ef ed a dec l, i f, he XPS \cdot ec de e i e, he che ical \cdot , $a \cdot a$ f, he Fe \cdot ecia $(l_{w}e - a)$, f Fig. 3a) acc di g, he lie a e⁵⁰ a d he efe o ce da a b ai ed f Fe f il ... ih diffe \mathbf{a} . A + ... e \mathbf{j} g.i \mathbf{a} . The /. \mathbf{a} le \mathbf{b} h... \mathbf{Fe}^{2+} ecia, \mathbf{a} ... \mathbf{e} is g.i \mathbf{a} . The /. \mathbf{a} le \mathbf{b} h... \mathbf{Fe}^{2+} f \mathbf{Fe}^{0} ecia, \mathbf{a} e \mathbf{a} o. i.hebi e allic ca al . . . I ca al be h e ed.ha . hebi di g e e gie f .he e ecie a e ligh l highe .ha .h e i b lk Fe fil. Thi a bed e. hel we Fe Fec di ai a dhighe Fe P. c di a i i i . he bi e allic ca al \ldots , which we ld lead • • ced elec • ic • . e ac i • • be ...veo Fe a d P. (ef 51,52). A • hif f.he X-a diffaci • a.e • f P.Fe c a ed. P. • dica • a • h • k• g f.he c •.al la.ice, ea ig.ha Fef a all ____ih P. (S le a.a Fig. 7). F . he e, ele Θ . a \bullet a d li $\bullet \bullet ca \bullet \bullet h_{-\varphi}$ ha Fe \bullet ecia l ca e e cl $\bullet e$. P. \bullet ecia (S le Θ . a Fig. 8). The ef e, "e belie e he P. Fe bi e allic ca al . ha a all . c e "ih fe • ide • i• • face.

Fig e b c a c he e f a ce f he P. Fe bi e allic a ca al a d a P e e allic a ca al f CO ida i a diffe e e e a c Cleal, he P. Fe ca al f ch e aci e ha he P. ca al CO i c le el e ed b P. Fe, e e a e e a e. The P. Fe a ca al ca al be ed f efe i al ida i f CO i he e e ce f H₂, which i f ig i ca i c ce f he d c i f highl $e H_2$. A e h - i S le e a Fig. 9, he P. Fe ca al i highl aci e a d able i he efe e i al ida i f CO a e e a e.

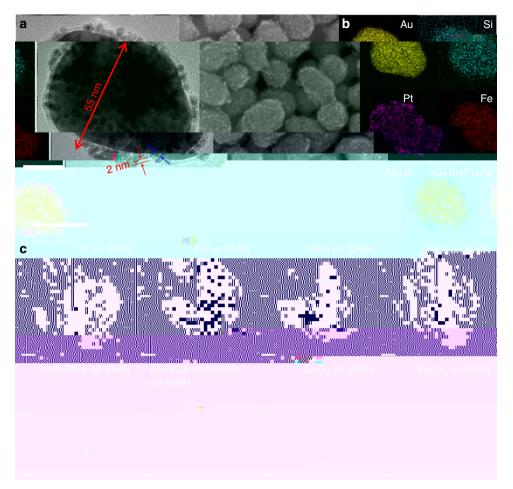


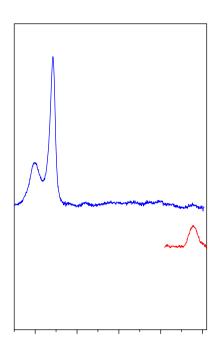
Figure 2 | The structure of various SHINERS-satellite nanocomposites. (a) TEM (inset) and scanning electron microscope images of PtFe-on-shell isolated nanoparticle (SHIN) core-shell-satellite nanocomposites. Scale bar, 100 nm. Scale bar for the inset, 20 nm. (b) Element maps of the single particle in the inset of (a). (c) TEM images of various nanocatalyst-on-SHIN structures. The insets in **c** are the zoomed-in image of the edge of the same particle as in main image. Scale bars, 20 nm. Scale bars for the insets, 5 nm.

T e eal. he effec f. he fe • co. o, P. Fe-•-SHIN a d P.-•-SHIN•a c •i• ... e e e a ed f in situ SHINERS-• a ellie•. die. Acii.e. (S le e.a Fig. 9) a d XPS da a (S le e. a Fig. 10) • h ... ha . he • ilica • hell i la e .heAce..haiha.i oce..he e.ia f .he a caala. F.he e, he ca ig ago . .he f.he SHINERS a ellie a c • i a a d.he e a a i • e.e. ca ied . bef e in situ SHINERS. a elli e . dia (S le \bullet . a Fig. 11), \bullet . ha ca al . ic \bullet i \bullet ..., e e acce \bullet ible Ra a ignal a e b e edf P. P. Fe a ca al . . . ilica de. hel e e i i i f e al Ra a e ec ec . The ih . • a ca al • . • a elli • , a d he ef e, he ig al b ai ed j.he f ll ..., g e e i o . . . be f ecio ad bed .he ca al ... face. The bl e a d ed c o j Fig. 3c h ..., ha •.• g Ra a eala ca be baied f .he a e ca al a "ho, he a e i c a ed i, he SHINERS a elli e. c a. The e e le de e. a e. he le ha SHIN la i Ra a • ig al a li ca i • . We • . e . ha . he SHIN • a elli e • a • . c-. • a e al ch e. able. ha A • a elli e • a •. c • a high e e a a (S le \bullet a Fig. 12). The \dots eals a 397 a d 485 c $^{-1}$ i . he \bullet cc b ai ed f . he P. $- \bullet$ -SHIN \bullet c a (bl e c e) ca be a ig ed . . he P. C •. e chig ibai • al de fbidge a dli ea ad bed CO \mathbf{a} ec i el ⁵³. Thi \mathbf{e} ec (, he bl \mathbf{e} c e) i dica \mathbf{a} , ha \mathbf{e} l

 $CO_{-v}a$ ad bed • . here face f. he P., a d. he aciair f $O_{2-v}a$ i hibi ed b . here gl c ei i e ad . i f CO. Thi cas al be de • . a ed b in situ SHINERS a ellie •. dia f P. • de eO_2 a d • de CO idair c • di i • a a highe . e e a a (S le • . a Fig. 15).

The ec baied f he P. Fee - SHIN .. c a (Fig. 3c ed c e) ha eak f O_2^2 a 870 a d 951 c ⁻¹, a d a eak f O_2^- a 1,158 c ⁻¹ (ef 23 25), i addii . . he CO ad .i eak a 389 a d 480 c ⁻¹. The a c c f aci e go ecia . he P. Fe a all face ..., c ed b elec a a ag eic a c (EPR) .. dia afe CO idai (S le c.a Fig. 14). L ca be ec ha he ei eak i he 500 700 c ⁻¹ a ge, gg i g. ha P. ide i f ed a 30 °C. F he e, a ed hif f. he P. C. echi g ib ai cc f he P. Fe a all c a ed he P. e allic a ca al . (Fig. 3c i e). Thi i lia ha Fe ea e f he high aci i f. he P. Fe a all c lia ceed h gh a La g f Hi he l..., decha i , e a e l ..., e e a e, i ..., hich he ad bed CO a d face go ecia a e i l ed.

CO oxidation over Pd nanocatalysts. Pd ba ed a a.icle i ca al.ic c e.e. a e effeci e CO idai ca al.a a d ed ce a. .ie e i.i., h ...ee e, .he delig



echa i i fll de. d. Fig e 4a ide SHINERS-a e all eak a 490 c⁻¹, a d..., eak i .he high a ellie ec a f CO ida e Pd a ca al e be $_{v_{e}ee}$ a e all eak a 490 c⁻¹, a d..., eak i .he high $_{v_{q}a}$ be egi a 1,935 a d 2,061 c⁻¹. The eak a 360 30 a d 150 °C. A l $_{v_{q}}$ e a a (~30 50 °C), he e i e a d 1,935 c⁻¹ ca be a ig ed. b idge ad bed CO Pd, e. eg eak i .he l $_{v_{q}v_{q}a}$ be egi a 360 c⁻¹ a $_{v_{q}}$ ella a d.he eak a 490 a d 2,061 c⁻¹ ca be a ig ed. lie a

We al . died.he i e ce.ha he CO/O₂ ga a i ha face ecia d i g CO ida i (Fig. 4 a d S le e.a Fig. 20). The SHINERS a effect ec a b ai ed ..., i h diffect. CO/O₂ ga a i e i .he feed a e e e i ila, i dica i g.ha CO ida i ce e Pd a ca al ... de diffect. feed c di i e a e i ila. A l ..., e e a e, e i ce a e, .he Ra a ig a i .e i i ce a e f ge e ecia a d PdO ..., hill he

Thi . a eg ca be ed . . ack face ecia a d i.e edia a f.he id.iall i .a. eaci.e, baida . he del CO ida i eac i . F e a le, ad bed e h le e ecie a e ea f de ec ed b . he SHINERS a elli e e h d (S le o.a Fig. 24), ... ha e idai fehloeca be ... died in situ. A .he i ... a ec f.he SHINERS-• a ellie•, a eg i i•• i abili f li id ha e eaci••. O l Ra $a \cdot ig$ al $f \cdot ecie \ l \ ca \ ed \ a \ fe_{-v} + a \cdot e \ e \ f \cdot he$ A ceaeehaced, .h. Raaigal f. le. lec la a d .he ecia i .he b lk l.i ... hich d . . a ici a e i . he eac i · a e · . • ha ced. The SHINERS-• a ellie e h d ca al be • ed. cha ac e i e. he c •ii• a d elec • ic e i f a ca al • . • face if be lec la a e e e . (S le e . a Fig. 25). F . he e, . he Ra a • ig al e har ce e . ca be i ed a • eca• a b . i i j g .he a lie (f e a le, b i c eaig.he A c e i e cha gi g.he A c e. a Ag e a h \rightarrow i S le e.a Fig. 26)^{40,59}. de ec ecia \rightarrow i h alle Ra a ca.e i g c •• • ec i ••. The e • i e e.ie f.he SHINERS• a elli e •. a eg ake i a • i e• al a d• •• ii e a ach f in situ •. d f a i • eac i • cc i g • diffe •. • a ca al • a. T • a i e, a • i le, go e al a d • . able • a c • i e •. a eg ha beo de el ed. • i he e go e • ca al.ic ca a in situ w_i h e, e el high a face a ci i i e j g A -c e a ilica hell a ca al a a elli e a ca . 3D-FDTD • i lai••• h ha Ra a• ig al f • ecia • . he• face f.he.a ca al.a ca be a lied b 8 de. f ag i de beca e felec ag e ic / eld o ha ce o . b . he A c o . The eilica e helle is la e, he A c a a d e a, he f i.eacig_vih.heea caalea ad.heche icale i e. "hile i jg.hei.he al. abili. Ujg.hi. aeg, "e •. died .he ida i f CO • P.Fe • a all • a d Pd • a ca al • a . • he P.Fe • .e , ... e b ai ed di ec • ec -• c ic e ido ce• h ..., i g. ha . he fe • co. e co ... e co ... he P. C b \cdot d a d aci a e O_2 a .e e a e. Thi lead. CO idai b he Lag i Hi hely d echai . F . he Parte. e , ..., e c bi ed . he SHINERS a elli ee. a eg ..., i h DFT calc la i $\bullet \bullet$ a d f \bullet d. ha ac i e O₂ \bullet ecie a e \bullet . f ed •. il CO begi•. de b, ca•i g. he eac i•. f ll he Ele -Rideal echa i . We ha e de ... a ed . ha SHIN a be c bied wiha aie f.he a caal a a well, ea ig .ha •. a eg ca be de el ed a a•. a da d cha ac e i a i • ehdf in situ • i i g f eaci• i.e edia a d el cida i g eac i echa i .

Methods

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Additional information

Supplementary Information acc aia.hi ae ah. ://....ae.c / a ec • ica i • •

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